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View Answer. Chemical equilibrium is established when the forward rate of reaction is equal to the reverse rate of reaction. a. TRUE b. FALSE. View Answer. For a reaction $N_2 + O_2 \rightarrow 2NO$. If $K_c...$

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Chemical equilibrium (definition) The state reached when the concentration of reactants and products remain constant over time What is the general, reversible reaction for equilibrium?

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Answer and Explanation: A chemical reaction is a dynamic equilibrium. In chemical reactions when the system is at equilibrium, reactions are still occurring. However, the rate of forward reaction...

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A chemical equilibrium is a dynamic equilibrium. True ...

Section 17.2 Factors Affecting Chemical Equilibrium. 1. If a stress is applied to a system at equilibrium, the system shifts in the direction that relieves the stress. 2. Answers may vary, but should include changes in concentration, volume, and temperature. Some students may list pressure in place of volume. 3. left; remove N_2 or H_2

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The equilibrium is represented by the

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equation above. (a) Write the expression for the solubility-product constant, K_{sp} , and calculate its value at 18°C . (b) Calculate the equilibrium concentration of Mg^{2+} in 1.000 liter of saturated MgF_2 solution at 18°C to which 0.100 mole of solid KF has been added. The KF dissolves completely.

AP Chemistry - Chapter 13, Chemical Equilibrium Study Guide

What is chemical equilibrium? A state of balance in a reaction where the forward and reverse reaction speed is unequal. A state where the reaction is unbalanced.

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